



Page: 1 of

Infosafe No™ VARAZ Issue Date : May 2021 ISSUED by HUNTERST

Product Name HYDROCHLORIC ACID

1. Identification

GHS Product

HYDROCHLORIC ACID

Identifier

Hunters Products (TAS) Pty. Ltd. (ABN 004 601 263) **Company Name**

60 Gleadow Street INVERMAY Address

TAS 7248 AUSTRALIA

Tel: 03 6331 4755 Telephone/Fax Fax: 03 6334 1065 Number 0407 610 542 **Emergency phone**

number

Recommended use of Precursor for generation of chlorine dioxide gas used in water treatment.

the chemical and restrictions on use

2. Hazard Identification

Skin Corrosion/Irritation: Category 1B GHS classification of

STOT Single Exposure: Category 3 (respiratory tract irritation) the

substance/mixture

DANGER Signal Word (s)

Hazard Statement (s) H314 Causes severe skin burns and eye damage.

H335 May cause respiratory irritation.

P102 Keep out of reach of children. Precautionary

statement - General P103 Read label before use. Corrosion, Exclamation mark Pictogram (s)





P260 Do not breathe dust/fume/gas/mist/vapours/spray. Precautionary P264 Wash contaminated skin thoroughly after handling. statement -P271 Use only outdoors or in a well-ventilated area. Prevention

P280 Wear protective gloves/protective clothing/eye protection/face

protection.

Precautionary statement - Response

P301+P330+P331 IF SWALLOWED: rinse mouth. Do NOT induce vomiting. P303+P361+P353 IF ON SKIN (or hair): Remove/Take off immediately all

contaminated clothing. Rinse skin with water/shower.

P304+P340 IF INHALED: Remove victim to fresh air and keep at rest in a

position comfortable for breathing.

P305+P351+P338 IF IN EYES: Rinse cautiously with water for several minutes.

Remove contact lenses, if present and easy to do. Continue rinsing.

P310 Immediately call a POISON CENTER or doctor/physician.

P363 Wash contaminated clothing before reuse.

P403+P233 Store in a well-ventilated place. Keep container tightly closed. **Precautionary**

P405 Store locked up. statement - Storage

P501 Dispose of contents/container in accordance with local regulations. **Precautionary**

statement - Disposal

3. Composition/information on ingredients

Ingredients	Name	CAS	Proportion	
	Hydrochloric acid	7647-01-0	>20%	
	Water	7732-18-5	to 100%	

4. First-aid measures

Inhalation Remove victim from area of exposure - avoid becoming a casualty. Remove

contaminated clothing and loosen remaining clothing. Allow patient to assume most comfortable position and keep warm. Keep at rest until fully recovered.

Print Date: 20/05/2021





5 Page: 2 of

Infosafe No™ VARAZ Issue Date : May 2021 ISSUED by HUNTERST

Product Name HYDROCHLORIC ACID

If patient finds breathing difficult and develops a bluish discolouration of the skin (which suggests a lack of oxygen in the blood - cyanosis), ensure airways are clear of any obstruction and have a qualified person give oxygen through a face mask. Apply artificial respiration if patient is not breathing.

Seek immediate medical advice.

Immediately rinse mouth with water. If swallowed, do NOT induce vomiting. Give Ingestion

a glass of water. Seek immediate medical assistance.

Skin If spilt on large areas of skin or hair, immediately drench with running water

and remove clothing. Continue to wash skin and hair with plenty of water (and

soap if material is insoluble) until advised to stop by the Poisons

Information Centre or a doctor.

If in eyes, hold eyelids apart and flush the eye continuously with running Eye contact water. Continue flushing until advised to stop by the Poisons Information

Centre or a doctor, or for at least 15 minutes. Continue to wash with large

amounts of water until medical help is available.

Eye wash station and normal washroom facilities. Emergency shower if handling First Aid Facilities

industrial quantities.

Product is a strong hydrochloric acid solution. If swallowed, vomiting should Advice to Doctor

not have been induced because of risk of aspiration of strongly acidic froth into the lungs. Toxic by inhalation. Causes severe burns. Contact Poisons

Information Centre.

5. Fire-fighting measures

Suitable extinguishing media Use extinguishing media appropriate to surrounding fire. Use water spray to

cool containers and surrounds.

Fire-fighters to wear self contained breathing apparatus and protective **Specific Methods**

equipment. If safe to do so remove containers from path of fire.

Specific hazards

arising from the

Liberates toxic fumes of hydrogen chloride in a fire. This material is incompatible with steel, other common metals and nylon. Alkalis may have

chemical

violent reactions.

Hazchem Code

6. Accidental release measures

Emergency

Dilute.

Procedures Spills & Disposal Increase ventilation.

For large spills:

Contain spillages with sand or earth. Transfer both liquid and solids to

suitable container(s). Treat residues as for small spills.

For small spills:

Spills may be neutralised by the liberal application of soda ash or crushed limestone. After reaction has ceased, mop up cautiously with plenty of water and run to waste, diluting greatly with running water. Otherwise, absorb on inert absorbent and transfer to suitable closed container. Wash site of spillage thoroughly with water and detergent. Ventilate area to dispel any

residual vapours.

7. Handling and storage

Precautions for Safe Handling

Avoid contact with skin and eyes. Avoid breathing concentrated vapours.

Conditions for safe storage, including any incompatibilities

Store in a cool, well ventilated place, out of reach of children. Large quantities should be stored in a bunded dangerous goods store. Store in original container. Keep container tightly closed and out of direct sunlight. Keep away from naked flames and other sources of ignition. Keep away from oxidising alkalis, oxidising agents and active metals. Protect from physical damage. Clean up all spills and splashes promptly; avoid secondary accidents. Incompatibles: Alkalis, other mineral acids, oxidising agents, active metals,

Unsuitable Materials

cyanides, sulphides, sulphites.

8. Exposure controls/personal protection

Occupational exposure limit values

STEL TWA Name





3 of 5 Page:

Infosafe No™ VARAZ Issue Date : May 2021 ISSUED by HUNTERST

Product Name HYDROCHLORIC ACID

mg/m3 mg/m3 Footnote ppmppmHydrochloric acid 7.5 5 Peak

limitation

Appropriate engineering controls Avoid using active metals (such as aluminium, tin, zinc, copper) as materials of construction. Ensure adequate ventilation (same as outdoors) when using. If handling industrial quantities, or if vapour risk exists, consider local $\hbox{\tt mechanical exhaust/extraction to keep airborne contamination as low as}$

possible and at least below the TLV.

Personal Protective Equipment

Avoid contact with skin and eyes. Avoid breathing vapours. Personal protection to be selected from those recommended below, as appropriate to mode of use,

quantity handled and degree of hazard:-

Normal Use:

Eye/face protection Gloves, rubber or plastic. Industrial Quantities:

Full face respirator fitted with acid vapour filters

Face shield or safety glasses Gloves, rubber or plastic Plastic apron, sleeves and boots

Impervious overalls.

Always maintain a high level of personal hygiene when using this product. That is wash hands before eating, drinking, smoking or using the toilet.

9. Physical and chemical properties

Liquid Form

Clear Colourless to Slightly Yellow liquid. **Appearance**

Characteristic pungent odour of hydrochloric acid. Odour

>98C **Boiling Point**

Miscible at all concentrations. Solubility in Water

Specific Gravity

Approx. 1.0 pН Not available Vapour Pressure

Flash Point None

Non flammable. Flammability

Other Information Reacts vigorously or violently with alkalis. Contact with carbonates or

bicarbonates will generate carbon dioxide, a simple asphyxiant. Contact with cyanides, sulphides or sulphites will generate very toxic gases. Corrosive to many common metals, generating hydrogen, a flammable gas. Corrosive to concrete floors and walls. May turn yellow on exposure to direct sunlight.

Slippery when spilled.

10. Stability and reactivity

Stable under normal use conditons. **Chemical Stability** Incompatible materials, sunlight. **Conditions to Avoid**

Incompatible Alkalis, oxidising agents, active metals, cyanides, sulphides, sulphites,

concrete. Materials

Hydrogen chloride, chlorine. Hazardous

Decomposition Products

Possibility of May react vigorously or violently with alkalis. Contact with carbonates or

bicarbonates generates carbon dioxide. hazardous reactions

11. Toxicological Information

Acute Toxicity - Oral LD50: Hydrochloric acid 900 mg/kg oral, rabbit.





5 Page: 4 of

Infosafe No™ VARAZ Issue Date : May 2021 ISSUED by HUNTERST

Product Name HYDROCHLORIC ACID

Corrosive. May be fatal. Will cause immediate pain, burns to the mouth, Ingestion

throat, oesophagus and gastrointestinal tract. May cause permanent tissue destruction of the oesophagus and digestive tract. Small quantities are likely

to cause gastric upset, nausea, vomiting and diarrhoea. An aspiration risk. Inhalation of vapours or aerosols may cause coughing, choking, inflammation of Inhalation

the nose, throat and upper respiratory tract, sore throat and shortness of breath. May cause tissue damage to the mucous membranes. Aspiration of acidic froth into the lungs during swallowing or vomiting may cause serious chemical pneumonitis (inflammation and damage to lung tissues) and pulmonary oedema

(fluid build-up in the lungs). Onset of symptoms may be delayed.

Corrosive. May cause redness, severe irritation and burns. Hydrochloric acid Skin

may be absorbed through the skin in harmful amounts. Will have a degreasing

effect on the skin.

Corrosive. May cause severe burns to eye tissues and permanent eye damage. Eye

Slight exposure may cause painful sensitisation to light. Over-exposure may

result in loss of sight.

Repeated or prolonged eye exposure to vapours may result in total loss of **Chronic Effects**

vision. Long term exposure to vapours may lead to erosion of the teeth.

Local concentrations may be harmful to aquatic organisms, including fish.

12. Ecological information

Harmful to aquatic organisms. **Ecotoxicity**

Readily transported by water. **Mobility**

Other Adverse

Effects

Environmental

Protection

Avoid contaminating waterways, drains, sewers, or ground.

13. Disposal considerations

Refer to appropriate authority in your State. Dispose of material through a Waste Disposal

licensed waste contractor. Normally suitable for disposal by approved waste

Special precautions

for landfill or incineration

Unsuitable for incineration. May be unsuitable for some landfill sites without

prior neutralisation.

Discharge of large quantities of acidic waste to concrete sewer may be **Local Legislation**

regulated by local authorities.

14. Transport information

Classified as a Class 8 Dangerous Good. Dangerous Goods of Class 8 Corrosives Transport

are incompatible in a placard load with any of the following: - Class 1, Class Information

4.3, Class 5, Class 6, if the Class 6 dangerous goods are cyanides and the

Class 8 dangerous goods are acids and Class 7.

U.N. Number 1789

UN proper shipping

name

HYDROCHLORIC ACID

Transport hazard

class(es)

Hazchem Code

2R

Packing Group

TT

EPG Number

8A1 40

IERG Number

15. Regulatory information

Poisons Schedule

All components listed. AICS (Australia)





5 of 5 Page:

Infosafe No™ VARAZ Issue Date : May 2021 ISSUED by HUNTERST

Product Name HYDROCHLORIC ACID

20/05/2021

16. Other Information

Date of preparation

or last revision of

SDS

Literature

Preparation of Safety Data Sheets for hazardous Chemicals Code of Practice

Standard for the Uniform Scheduling of Medicines and Poisons References

Australian Code for the Transport of Dangerous Goods by Road & Rail Globally Harmonised System of classification and labelling of chemicals

Signature of Preparer/Data

Service

Technical Contact

Emergency Advice All Hours:

Technical Manager 0407 610 542

Numbers

Technical Manager: 0407 610 542 Mon-Fri 8am - 6pm Poisons Information Centre: 13 11 26 - 24hrs

Transport/Fire Emergency: 000 (Emergency services)

Other Information

This SDS summarises at the date of issue our best knowledge of the health and safety hazard information of the product, and in particular how to safely handle and use the product in the Workplace. Please refer to the technical datasheet (Instructions for use), and the label on the drum. The company cannot anticipate or control the individual working conditions encountered and so each user should read this SDS carefully, and if in doubt ring the Contact Point Number given below.

...End Of MSDS...

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