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Infosafe No™ VARFF Issue Date : November 2020 ISSUED by HUNTERST

Product Name SODIUM HYPOCHLORITE 12.5%

1. Identification

GHS Product

SODIUM HYPOCHLORITE 12.5%

Identifier

Hunters Products (TAS) Pty. Ltd. (ABN 004 601 263) **Company Name**

60 Gleadow Street INVERMAY Address

TAS 7248 AUSTRALIA

Tel: 03 6331 4755 Telephone/Fax Fax: 03 6334 1065 Number 0407 610 542 **Emergency phone**

number

the chemical and

Recommended use of As a chlorinating bleach and sanitiser. Use as directed on the product label. In prescribed premises, rinse all food areas with potable water after use.

restrictions on use

2. Hazard Identification

GHS classification of CM1 Corrosive to Metals: Category 1

ED1 Eye Damage/Irritation: Category 1 the

SC-1B Skin Corrosion/Irritation: Category 1B substance/mixture

DANGER Signal Word (s)

Hazard Statement (s) AUH031 Contact with acids liberates toxic gas.

H290 May be corrosive to metals.

H314 Causes severe skin burns and eye damage.

H318 Causes serious eye damage.

P101 If medical advice is needed, have product container or label at hand. **Precautionary**

P102 Keep out of reach of children. statement - General

P103 Read label before use.

Corrosion Pictogram (s)



P260 Do not breathe dust/fume/gas/mist/vapours/spray. **Precautionary**

P264 Wash contaminated skin thoroughly after handling. statement -

P280 Wear protective gloves/protective clothing/eye protection/face

protection.

Precautionary

Prevention

statement – Response

P301+P330+P331 IF SWALLOWED: rinse mouth. Do NOT induce vomiting. P303+P361+P353 IF ON SKIN (or hair): Remove/Take off immediately all

contaminated clothing. Rinse skin with water/shower.

P304+P340 IF INHALED: Remove victim to fresh air and keep at rest in a

position comfortable for breathing.

P305+P351+P338 IF IN EYES: Rinse cautiously with water for several minutes.

Remove contact lenses, if present and easy to do. Continue rinsing.

P310 Immediately call a POISON CENTER or doctor/physician.

P363 Wash contaminated clothing before reuse.

P390 Absorb spillage to prevent material damage.

statement - Storage

P405 Store locked up. **Precautionary**

P501 Dispose of contents/container in accordance with local regulations. **Precautionary**

statement - Disposal

3. Composition/information on ingredients

Ingredients	Name	CAS	Proportion
	Sodium Hypochlorite 12.5%	7681-52-9	100%

4. First-aid measures





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Remove from exposure, rest and keep warm. In severe cases, obtain medical Inhalation

attention. Apply artificial respiration if not breathing.

Immediately rinse mouth with water. Do NOT induce vomiting. Slowly give water Ingestion

to drink. Seek medical assistance.

If skin contact occurs, remove contaminated clothing and wash skin thoroughly. Skin

If swelling, redness, blistering, or irritation occurs seek medical advice.

If in eyes, hold eyes upen, flood with water for at least 15 minutes and see a Eye contact

Eye wash station and normal washroom facilities. Emergency shower if handling **First Aid Facilities**

industrial quantities.

Product is a solution of sodium hypochlorite. If swallowed, may lead to fall Advice to Doctor

> in blood pressure. Treat with antacids to neutralise hypochlorous acid formed in the stomach, then as for alkaline materials. Onset of pulmonary oedema, following inhalation overexposure, may be delayed. Treat symptomatically.

Contact Poisons Information Centre.

5. Fire-fighting measures

Use extinguishing media appropriate to surrounding fire. Suitable

extinguishing media

Hazards from

Chlorine, hydrogen chloride.

Combustion **Products Specific Methods**

In case of small fire/explosion use water. In case of major emergency use PPE:

breathing apparatus and protective gloves.

Specific hazards arising from the chemical

Contact with combustible material may cause fire. May form explosive products with primary aliphatic or aromatic amines, methanol and with nitrites. Contact

with acids will generate chlorine, a toxic and corrosive gas. May react vigorously or violently with oxidising agents, reducing agents and metal

salts.

Hazchem Code

Avoid contact with coloured fabric as Chlorine may bleach colour out. Other Information

May give off dangerous gas if mixed with other products.

6. Accidental release measures

Contain. **Emergency**

Increase ventilation. **Procedures** For large spills: Spills & Disposal

Contain spillages with sand or earth. Transfer both liquid and solids to

suitable container(s). Treat residues as for small spills.

For small spills:

If local regulations permit, mop up with plenty of water and run to waste, diluting greatly with running water. Otherwise absorb on inert absorbent and transfer to suitable closed container. Wash site of spillage thoroughly with

water and detergent. Ventilate area to dispel any residual vapours.

7. Handling and storage

Precautions for Safe Handling

Avoid contact with skin and eyes. Avoid breathing concentrated vapours. Keep away from combustible materials, acids, nitro compounds including amines and

ammonium compounds, other chlorinating materials.

Conditions for safe storage, including any incompatibilities

Store in a cool, well ventilated place, out of reach of children. Large quantities should be stored in a bunded dangerous goods store. Store in original container. Containers should be vented with at least 5 % ullage, or unvented with at least 10 % ullage @ 23 $^{\circ}\text{C}.$ Do not store in unlined metal drums. Keep container tightly closed and out of direct sunlight. Prevent vapours from collecting in enclosed or low lying spaces. Keep away from acids, other oxidising agents, combustible materials. Protect from physical damage.

Clean up all spills and splashes promptly; avoid secondary accidents. Unsuitable Materials Incompatibles: Acids, other oxidising agents, combustible materials, metals,

metal salts, formic acid, amines, ammonium compounds, aziridine, methanol.

8. Exposure controls/personal protection





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Appropriate

Avoid using wood, wood products or unprotected metals as materials of engineering controls

construction. Ensure adequate ventilation (same as outdoors) when using. If handling industrial quantities, or if vapour risk exists, consider local mechanical exhaust/extraction to keep airborne contamination as low as

possible and at least below the TLV.

Personal Protective Equipment

Prevent contact with the eyes. Avoid contact with the skin. Avoid breathing the vapours. Personal protection to be selected from those recommended below,

as appropriate to mode of use, quantity handled and degree of hazard:-

Normal Use:

Eye/face protection Gloves, rubber or plastic. Industrial Quantities:

Face shield or safety glasses Gloves, rubber or plastic Plastic apron, sleeves and boots

Impervious overalls.

Always maintain a high level of personal hygiene when using cleaning chemicals. That is wash hands before eating, drinking, smoking or using the

toilet.

CAUTION: Cotton or linen overalls impregnated with oxidising agents may be

readily ignited and can burn fiercely.

9. Physical and chemical properties

Liquid **Form**

Clear yellow liquid **Appearance** Clear pale yellow. Colour

Odour Strong smell of chlorine.

100C **Boiling Point**

Miscible with water in all proportions. Solubility in Water

Specific Gravity 1.2

12.0 - 13.0 pН Vapour Pressure Not available.

None **Flash Point**

Flammability Not flammable. Moderate oxidiser. Contact with combustible materials may cause

fire.

Oxidiser. Contact with combustible material may cause fire. Contact with acids Other Information

will generate chlorine, a toxic and corrosive gas. May react violently with reducing agents. Can react with primary aliphatic and aromatic amines, methanol and nitrites to give explosive products. May react vigorously with oxidising agents. Incompatible with most metals. Will decompose on standing, generating chlorine. Decomposition will be accelerated by contamination and by exposure to sun light. May react vigorously with peroxides and metal salts. On

long storage, may generate pressure inside sealed containers.

cautiously.

10. Stability and reactivity

Incompatible materials, heating, sunlight, prolonged exposure to air. **Conditions to Avoid**

Acids, other oxidising agents, combustible materials, amines, ammonium Incompatible

compounds, formic acid, methanol, phenylacetonitrile, reducing agents, metal Materials

salts, metals, wood and wood products.

Chlorine, hydrogen chloride. Hazardous

Decomposition Products

Contact with combustible materials may cause fire. May react explosively with Possibility of

formic acid, phenylacetonitrile. Forms explosive products with amines, hazardous reactions ammonium compounds, aziridine, methanol. Contact with acids will generate

chlorine gas.





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11. Toxicological Information

Acute Toxicity - Oral Acute oral toxicity (LD50): 5800 mg/kg [Mouse]

Acute Toxicity -

LCLo Chlorine 2,530 mg/m3/30 minutes human

Inhalation

Ingestion Corrosive and irritating if swallowed. May cause corrosion of the mucous

membranes of the mouth, throat and gastrointestinal tract, a burning sensation, pain, abdominal cramps, oesophageal or gastric perforation, laryngeal oedema. May cause general depressed activity, lowering of blood pressure, nausea, vomiting, weakness, delirium, loss of consciousness and

coma.

Inhalation Inhalation of vapours or aerosols may cause coughing, shortness of breath and

> a burning sensation. May cause severe bronchial irritation and pulmonary oedema (fluid build-up in the lungs). Onset of symptoms may be delayed by

several hours after exposure.

Pulmonary complications, e.g. from aspiration, may contribute to the death of

a casualty.

Skin Short contact may cause irritation. On longer contact risk of chemical burns.

Corrosive or irritating to eyes. May cause redness, pain, severe deep burns. Eye

Prolonged contact may lead to permanent injury.

Chronic skin exposure may cause skin sensitisation in some rare cases, but **Chronic Effects**

sodium hypochlorite is not classified as a sensitiser. Chronic exposure to sodium hypochlorite may lead to methaemoglobinaemia, characterised by chocolate-brown coloured blood, headache, dizziness, weakness, shortness of breath, cyanosis, rapid heart rate, unconsciousness and possible death.

12. Ecological information

Ecotoxicity Harmful to aquatic organisms.

Persistence and

Will degrade on exposure to air, sunlight.

degradability

Mobility

Readily transported by water and through soil.

Environmental Fate Decomposes over time.

Environmental Protection

Avoid contaminating waterways, drains, sewers, or ground.

13. Disposal considerations

Waste Disposal

Do not discharge effluent containing this product into laes, streams, ponds, estuaries, oceans, or public waters unless this product is specifically

identified and addressed in a permit. Do not discharge effluent containing

this product without previously notifying the sewage treatment plant

authority. For guidance, contact your State Water Board or Regional Office of the EPA. Reduce with agents such as bisulfites or ferrous salt solutions. Some heat will be produced. Keep on alkaline side and dilute with copious amounts of water. The main end-product is salt water. Waste must be disposed of in accordance with federal, state and local environmental control regulations.

Refer to Land Waste Management Authority in your State.

Special precautions for landfill or

Unsuitable for incineration. May be unsuitable for landfill.

incineration

14. Transport information

Classified as a Class 8 Dangerous Good. Dangerous Goods of Class 8 Corrosives **Transport** are incompatible in a placard load with any of the following: - Class 1, Class Information

4.3, Class 5, Class 6, if the Class 6 dangerous goods are cyanides and the Class 8 dangerous goods are acids and Class 7. Store away from acids.

1791 U.N. Number

UN proper shipping

HYPOCHLORITE SOLUTION

name





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Transport hazard

class(es)

2X **Hazchem Code** III **Packing Group EPG Number** 8A1 **IERG Number** 37

15. Regulatory information

S5 **Poisons Schedule**

All components listed. AICS (Australia)

26/11/2020

16. Other Information

Date of preparation

or last revision of

SDS

Literature References Preparation of Safety Data Sheets for hazardous Chemicals Code of Practice

Standard for the Uniform Scheduling of Medicines and Poisons

Australian Code for the Transport of Dangerous Goods by Road & Rail Globally Harmonised System of classification and labelling of chemicals

Signature of Preparer/Data

Service **Technical Contact**

Emergency Advice All Hours:

Technical Manager 0407 610 542

Numbers

Technical Manager: 0407 610 542 Mon-Fri 8am - 6pm Poisons Information Centre: 13 11 26 - 24hrs Transport/Fire Emergency: 000 (Emergency services)

Other Information

This SDS summarises at the date of issue our best knowledge of the health and safety hazard information of the product, and in particular how to safely handle and use the product in the Workplace. Please refer to the technical datasheet (Instructions for use), and the label on the drum. The company cannot anticipate or control the individual working conditions encountered and so each user should read this SDS carefully, and if in

doubt ring the Contact Point Number given below.

...End Of MSDS...

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